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VERBS\_SP\_CLIENT@ClientName, @ClientId, @ClientEmail The problem with this scenario is when there is a procedure with the name ClientName\_proc, I will not be able to get its values from main procedure. I can, however, get the values I need with ClientId\_proc and ClientEmail\_proc with no

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A: Windows XP SP3 with Service Pack 2 (SP2) does not need to be activated. The reason they gave you about not being able to activate it on a retail machine is because Windows 7 and later requires a serial number to activate. There's nothing stopping you from activating this Windows XP SP3, you just have to use a legit serial number. The reason the download is not working for you is because that file has not been released by Microsoft yet, it is still being tested and final version of the file will be released in the near future. A process for preparing optically active compounds is known in which, for example, an optically active compound is adsorbed to an inorganic carrier such as talc, and then said compound is recovered from the inorganic carrier, for example, by the use of a solvent. As to this process, in order to recover the optically active compound from the inorganic carrier, the process must be performed so that the amount of the optically active compound adsorbed to the carrier is not lost. For example, the following processes may be considered as a process for making a recovery of the optically active compound. (1) Adsorption of optically active compound to a large amount of inorganic carrier is made, followed by reducing the amount of the adsorbed compound with a reducing agent to cause the adsorbed compound to be desorbed. (2) Adsorption of the optically active compound to the inorganic carrier is made, followed by increasing the amount of the adsorbed compound with a solvent to cause the adsorbed compound to be desorbed. In the processes, however, the amount of the adsorbed compound which can be used is limited by the amount of the carrier. Especially, in the process (2), since the amount of the adsorbed compound which can be used is limited by the adsorbing capacity of the inorganic carrier, the reaction rate is low, and further, the amount of recovered optically active compound is small. In addition, since the optically active compound is recovered in a liquid form, the obtained optically active compound is liable to be purified by distillation. In addition, a process of recovering an optically active compound from an inorganic carrier is known which is performed by heat treating an optically active compound adsorbed to the inorganic carrier (see, for example, Patent Document 1). In this process,